

























• Can also calculate kinetic energy by differentiating with respect to the mass $K = -\frac{mdZ}{bZdm}$

• Or use the "direct" form:
$$K = \left\langle e^{S}(-I\nabla^{2})e^{-S}\right\rangle_{pat}$$

• For pressure, differentiate wrt the volume (virial estimator).

$$P = \frac{1}{3V} \left[2T - \frac{1}{t} \sum_{i < j} \left\langle \mathbf{r}_{ij} \nabla u(r_{ij}) \right\rangle \right]$$

- In general, one can have different "estimators" having different convergence of systematic (Trotter) or statistical errors.
- Statistical errors require careful estimation.
- Other errors can be bias and finite-size errors.
- Free energy calculated just as in classical simulation, with all the same problems.

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	Dictionary of the Isomo	Quantum-Classical orphism
Prop p Atte	perties of a quantum syst properties of the fictitious ention: <i>some words have op</i>	em are mapped into classical polymer system posite meanings.
	Quantum	Classical
	Bose condensation	Delocalization of ends
	Boson statistics	Joining of polymers
	Exchange frequency	Free energy to link polymers
	Free energy	Free energy
	Imaginary velocity	Bond vector
	Kinetic energy	Negative spring energy
	Momentum distribution	FT of end-end distribution
	Particle	Ring polymer
	Potential energy	Iso-time potential
	Superfluid state	Macroscopic polymer
	Introlemand	Polymer length





































Special Potentials Coulomb Hard Sphere Coulomb: eigenfunctions Expansion in partial waves • are hydrogen atom simple: spherical bessel wavefunctions and functions+phase shifts hypergeometric function Various analytic approximations lots of analytic formulas, asymptotic formulas. Harmonic Oscillator · Can use super-symmetry to get rid of one variable: simplifies making tables. First rotate to diagonal Gets rid of the infinity in the representation to get a product attractive Coulomb of 1D density matrices singularity. · Can do analytically Describes hydrogen atom exactly.

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